

Crystallographic Structure Determination of the Diterpenoid Baccatin-V, a Naturally Occurring Oxetan with a Taxane Skeleton

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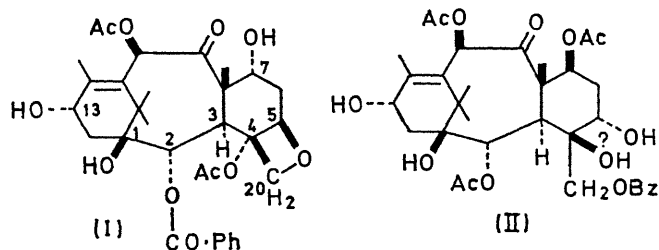
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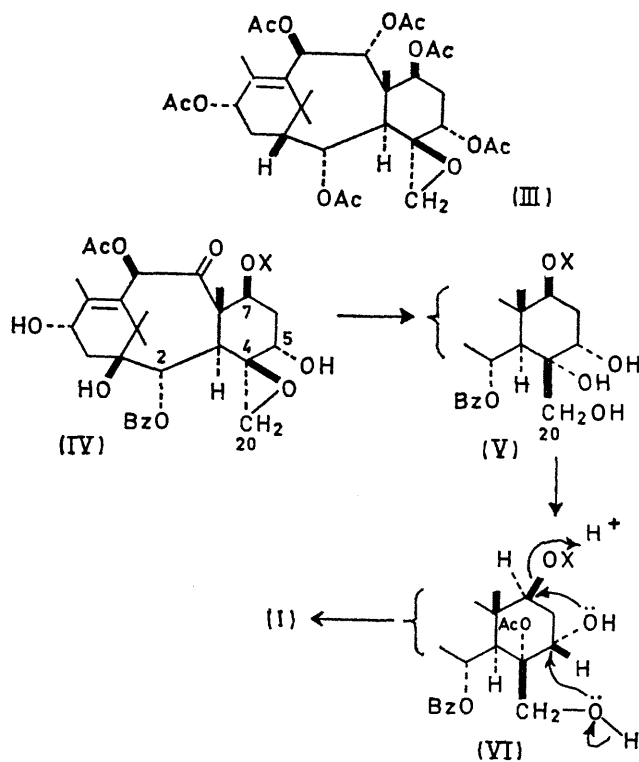
Summary Baccatin-V, $C_{31}H_{38}O_{11}$, a new diterpenoid from *Taxus baccata* L. has been shown by X-ray analysis to contain an oxetan ring and to be 5 β ,20-epoxy-1 β ,2 α ,4 α ,7 α ,10 β ,13 α -hexahydroxytax-11-en-9-one 4 α ,10 β -diacetate 2 α -benzoate (I).

A NEW compound, baccatin-V, $C_{31}H_{38}O_{11}$, m.p. 254–255°, $[\alpha]_D^{25} - 87^\circ$, has been isolated from *Taxus baccata* L. It occurs in fractions of extract near to those containing baccatin-III (II),^{1,2} which also has a similar negative rotation, and baccatin-IV.¹ The n.m.r. spectrum indicated two acetate groups and a benzoate group and had four singlets at τ 8.96, 8.90, 8.38, and 8.01 typical of the methyl groups of a taxane derivative. Its low-field region was similar to that of the spectrum of baccatin-III. A singlet at τ 3.17



indicated a proton at C-10 in the grouping $\cdot\text{CHOAc}\cdot\text{CO}\cdot$, the carbonyl group being at C-9. Hydroxy-groups were shown by ν_{max} at 3600 cm^{-1} . One was readily acetylated, when a signal at τ 5.1 shifted to τ 3.82 (broad t, J 8 Hz). This showed the probable presence of a 13 α -hydroxy-group.

An insufficient amount of baccatin-V was available for detailed chemical study, so a structure analysis was attempted directly on baccatin-V.



SCHEME

Baccatin-V crystallised in the non-centrosymmetric orthorhombic space group $P2_12_12_1$ with $a = 25.75$, $b = 9.31$, $c = 12.34$ Å, and $Z = 4$ for $C_{31}H_{38}O_{11}$. 3364 reflections

were collected on a Hilger-Watts linear diffractometer with Mo radiation, 2462 of which were taken as significantly above background. A computer application of the tangent formula of Karle and Hauptman³ was used to solve the structure. Eight phases were used as a starting set, four of which were fixed⁴ and four given 64 different combinations of trial values. After extension of each trial to about 220 phases, one set had a considerably lower "R value"⁵ and greater number of interactions in use. An *E* map calculated on 295 reflections phased by extending this set revealed practically the whole structure, which was completed from an *F* (obs) map phased on 37 correct, and one incorrect, positions. The structure of baccatin-V so derived is (I). The present value of *R* is 10.6%.

The structure is interesting in two respects. First it is a naturally occurring oxetan. Secondly the configurations of the oxygen atoms attached to C-5 and C-7 (β and α , respectively) are *opposite* to those found in all other taxane derivatives (*cf.* ref. 6). The reason for this may be associated with the biogenesis of the oxetan ring. This could involve a precursor such as the 4 β -20-epoxide (IV) in a sequence of interactions as exemplified in the Scheme. A taxane-4 β ,20-epoxide, baccatin-I,⁷ (III) is already known. The conversion of (VI) into (I) could be acid-catalysed or could be initiated by loss of XO⁻.

Oxetan formation by the mechanism suggested may be facilitated by compression between C-20 and the C-2-oxygen.

In (I) the interatomic distance between these atoms is only 2.6₈ Å. The C-2-oxygen is also very close (2.6₀ Å) to that at C-1. The importance of release of non-bonding strain in the oxetan formation has been discussed in the case of 3 α ,5 α -epoxycholestane.⁸

The n.m.r. signal of the β -proton at C-7 appears at τ 6.32 as a sextet being split by the C-6 methylene group and the hydroxylic proton of the 7 α -hydroxyl. On deuterium exchange, the signal collapses to a triplet. This hydroxylic proton is involved in a strong hydrogen bond with the acyl oxygen of the 4 α -acetate group, the distance between this oxygen and that of the 7 α -hydroxyl being 2.7, Å. The 2 α -proton gives a doublet at τ 4.26, being coupled with the 3-proton resonating at τ 5.98 (*J* 6 Hz). This latter proton is very near the double bond, being separated from it by only 2.7–2.8 Å. The 3-proton is enclosed within a globular-shaped carbon framework.

The methylene group of the oxetan ring resonates at τ 5.62 as a singlet in baccatin-V and its 13 α -acetate, but in the hexa-hydro-derivative of the acetate (benzoate group reduced) it is split into an AB system at τ 5.43 and 5.64 (*J* 6 Hz). The C-5 oxetan proton gives a signal at τ 5.1 (m).

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¹ W. R. Chan, T. G. Halsall, G. M. Hornby, A. W. Oxford, W. Sabel, K. Bjåmer, G. Ferguson, and J. W. Robertson, *Chem. Comm.*, 1966, 923.

² D. P. Della Casa de Marcano, T. G. Halsall, and G. M. Hornby, *Chem. Comm.*, 1970, 216.

³ J. Karle and H. Hauptman, *Acta Cryst.*, 1956, **9**, 635.

⁴ J. Karle and H. Hauptman, *Acta Cryst.*, 1956, **9**, 45.

⁵ J. Karle and I. L. Karle, *Acta Cryst.*, 1966, **21**, 849.

⁶ D. P. Della Casa de Marcano and T. G. Halsall, *Chem. Comm.*, 1969, 1282.

⁷ D. P. Della Casa de Marcano and T. G. Halsall, *Chem. Comm.*, preceding communication.

⁸ R. B. Clayton, H. R. Henbest, and M. Smith, *J. Chem. Soc.*, 1957, 1982.